





DEPARTMENT OF PURE & APPLIED CHEMISTRY

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TABLE OF SPECIFICATIONS

Chem 21 - General Chemistry II Laboratory

2nd Semester AY 2021-2022

Examination: ___ Midterm __ Final __ Long Exam

Date of Examination: June 23, 2022

	1				Taxon	omy of	Object	ives		
Content	No. of Meeti	Course Outcome/L earning Outcome	%	Remembering	Understanding	Applying	Analyzing	Evaluating	Creating	Total Items
	ngs	(CO/LO)		31.03	18.97	27.59	20.69	1.72	udies us	
				%	%	%	%	%		
Factors Affecting Rates of Chemical Reactions Chemical Equilibrium	1	CO1: Explain the theories and principles of chemical equilibrium and how these apply to the physical world	10.34	3	1	2	1			5
•		CO2: Apply								
Measurement of the Solubility Product Constant for a	1	safety precautions in the laboratory.	15.52	2	2	2	2			9
Slightly Soluble Salt		CO3: Analyze and present experimental data.								
Electrolytes: Acids, Bases, and Salts	1		12.07	3	1	2				7
pH Determination	1		17.24	3	2	3	2			10
Relative	1									
Strength of Acids and Bases	1		8.62	1	1	2				5
Common-lon Effect & Buffers	1		13.79	2	1	2	1			8
Spot Tests for Some Common lons	1		12.07	1		1		1		7
Total	8		100 %	18	9	16	6	1		50

Item Arrangeme	I. 1-11 II. 1-8	I. 12- 15 V. 1-5	III. 1- 16	IV. 1-6	I. 16	
nt						

Type/s of Test:

Test I

Multiple Choice Matching Type Test IV Calculations Test V Essay

Test III

Modified True or False

	Name of Course Instructor /Professor	Signature	Date Signed
Prepared by:	MARK RYAN R. TRIPOLE	A.	6/22/2

Department Instructional Materials Review Committee:

Committee	Name	Signature	Date Signed
Member:	ATOZ A. VASQUEZ	Garage	C(23) 22
Member:	MA. ROBELYN A. INSIK	mandel	6/23 /22
Chairperson	JACOB GLENN F. JANSALIN	fann	apspor
Department Head:	ELIZABETH S. QUEVEDO	O Where	6/23/20m

	Name	Signature	Date Signed
Verified by:	MA. THERESA P. LORETO Dean, CAS		
Validated by:	NANCY D. ABUNDA Head, OIMD	Ta :	

Note: A copy of the test paper with answer key shall be attached to the TOS.

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