



TABLE OF SPECIFICATIONS
Chem 157.1 – Physical Chemistry Laboratory I
2nd Semester AY 2021-2022

Examination: ___ Midterm ☒ Final ___ Long Exam

Date of Examination: June 23, 2022

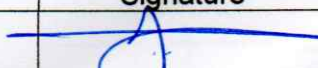
Content	No. of Meetings	Course Outcome/Learning Outcome (CO/LO)	%	Taxonomy of Objectives						Total Items
				Remembering	Understanding	Applying	Analyzing	Evaluating	Creating	
				31.03 %	18.97 %	27.59 %	20.69 %	1.72 %		
Ideal Gas Law in Food	1	CO1: Explain the theories and principles of physical chemistry behind the experimental procedures involved.	10.34	4	1		1			6
Real Gases	1		10.34	2	1	2	1			6
Determination of Calorific Value in Food Samples Using Bomb Calorimetry	1		15.52	3	2	3	1			9
Determination of Partition Coefficient	1	CO2: Apply safety precautions in the laboratory.	12.07	3	1	2	1			7
Distillation of Binary Liquids	1		17.24	3	3	4				10
Steam Distillation	1		8.62	1	2	2				5
Phase Diagram in a Three-Component Liquid System	1	CO3: Analyze and present experimental data.	13.79	1		1	6			8
Molar Mass Determination by Freezing Point Depression	1		12.07	1	1	2	2	1		7
Total	8		100 %	18	11	16	12	1		58

Item Arrangement				I. 1-11 II. 1-8	I. 12-17 V. 1-5	III. 1-15	IV. 1-12	I. 18		
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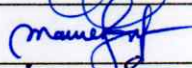
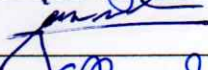

Type/s of Test:

Test I Multiple Choice
 Test II Matching Type
 Test III Modified True or False

Test IV Calculations
 Test V Essay

Prepared by:	Name of Course Instructor /Professor	Signature	Date Signed
	MARK RYAN R. TRIPOLE		6/22/22

Department Instructional Materials Review Committee:

Committee	Name	Signature	Date Signed
Member:	ATOZ A. VASQUEZ		6/23/22
Member:	MA. ROBELYN A. INSIK		6/23/22
Chairperson	JACOB GLENN F. JANSALIN		6/23/2022
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	Name	Signature	Date Signed
Verified by:	MA. THERESA P. LORETO Dean, CAS		
Validated by:	NANCY D. ABUNDA Head, OIMD		

Note: A copy of the test paper with answer key shall be attached to the TOS.

(3) Distribution of copies: OHIMD, Department, Faculty

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